Mass Cytometry

DOI: 10.1002/anie.201405233

uses stable isotopes with high atomic mass as tags and

harnesses the sensitivity, dynamic range, and single mass unit

resolution of inductively coupled plasma mass spectrometry

to allow for highly parameterized experiments (theoretically over 100 parameters). [2-4] This technology has been success-

fully used to monitor differential responses across the human

hematopoietic continuum based on 34 unique parameters.^[4]

Recent work has adapted MC-compatible reagents for 32-

parameter imaging of breast cancer tissue, revealing spatial

variations in tumor heterogeneity with an unprecedented

reagents bearing unique heavy isotopes (≥ 100 amu) with

a low biological background that report on meaningful

biological parameters. To date, the most successful reagents

developed have been bifunctional polymers that chelate

lanthanide ions and allow for antibody conjugation (MaxPar

Reagents, Figure 1 a). [6] Additionally, metal chelates that bind

or may be incorporated into DNA have been exploited (Figure 1b).^[7] The success that researchers have had with

these reagents cannot be understated; however, the potential

of MC should not be limited to reagents that measure static

biomarkers. Probes for specific enzyme activities would

To fully realize the parameterization of MC requires many

Identification of Hypoxic Cells Using an Organotellurium Tag **Compatible with Mass Cytometry****

Landon J. Edgar, Ravi N. Vellanki, Adrienne Halupa, David Hedley, Bradly G. Wouters, and Mark Nitz*

level of detail.^[5]

Abstract: Mass cytometry (MC) offers unparalleled potential for the development of highly parameterized assays for characterization of single cells within heterogeneous populations. Current reagents compatible with MC analysis employ antibody-metal-chelating polymer conjugates to report on the presence of biomarkers. Here, we expand the utility of MC by developing the first activity-based probe designed specifically for use with the technology. A compact MC-detectable telluroether is linked to a bioreductively sensitive 2-nitroimidazole scaffold, thereby generating a probe sensitive to cellular hypoxia. The probe exhibits low toxicity and is able to selectively label O₂-deprived cells. A proof-of-concept experiment employing metal-bound DNA intercalators demonstrates that a heterogeneous mixture of cells with differential exposure to O2 can be effectively discriminated by the quantity of tellurium-labeling. The organotellurium reagents described herein provide a general approach to the development of a large toolkit of MC-compatible probes for activity-based profiling of single cells.

Multiparameter analysis of single cells has traditionally been performed using fluorescence-based flow cytometry (FBFC); a technique that has transformed understanding of cellular heterogeneity. As scientists begin to ask increasingly complex questions about relationships between large numbers of biological parameters, the limitations inherent to FBFC have become apparent. Specifically, the number of fluorophore tags that may be employed simultaneously in any given FBFC experiment is limited by spectral overlap, thereby prohibiting a highly parameterized analysis.^[1] The introduction of mass cytometry (MC) has provided a means for overcoming this limitation. In place of fluorophores, MC

unlock new dimensions to MC analysis (Figure 1c). Here, **MaxPar Reagents** 1000s of kDa

> c) **Proposed Activity-Based Probe**



Figure 1. a) Generalized structure of commercially available bifunctional lanthanide-chelating polymer-antibody conjugates (MaxPar reagents). b) Structures of the heavy-isotope-containing nucleic acid intercalators employed in this study and the unnatural nucleoside analogue 5-iododeoxyuridine (IdU). c) Requirements and generalized design of a MC-compatible probe for studying dynamic/active biochemical processes.

Dr. R. N. Vellanki, Prof. D. Hedley, Prof. B. G. Wouters Princess Margaret Cancer Centre, Departments of Radiation Oncology and Medical Biophysics, University Health Network 101 College Street, Toronto, Ontario, M5G 1L7 (Canada)

[**] We acknowledge Tina Chen for collecting cytometry data, the U of T NMR facility for spectral analysis (CFI/ORF no 19119), the AIMS laboratory for mass spectrometric analysis, Dr. R. Williams, Dr. L. Willis, and Dr. Q. Chang for productive discussions and funding from the Natural Science and Engineering Research Council (NSERC), DVS Sciences, and the Canadian Cancer Society (no



Supporting information for this article is available on the WWW under http://dx.doi.org/10.1002/anie.201405233.

Nucleic Acid

Interrogators

^[*] L. J. Edgar, Dr. A. Halupa, Prof. M. Nitz Department of Chemistry, University of Toronto 80 St. George Street, Toronto, Ontario, M5S 3H6 (Canada) E-mail: mnitz@chem.utoronto.ca

we introduce a new approach for the development of activity-based probes compatible with MC technology using a compact telluroether scaffold.

Tellurium provides favorable characteristics for the development of small MC probes. The element forms telluroether and tellurophene functionalities which are small and lipophilic; properties which we hypothesize will minimize perturbation of attached biologically sensitive functionalities (i.e. activity-based groups), thus preserving their biological activity. This is in direct contrast to the lanthanide-chelating polymers used as mass tags in MaxPar reagents, because these groups are large and polar (Figure 1a). Additionally, five stable tellurium isotopes are available commercially, thereby allowing for multiple MC-distinguishable probes to be generated using identical chemistry. As a proof of concept of such a probe, we have targeted cellular hypoxia due to its importance in tumor biology and the well-defined chemical methods for interrogation.

The development of hypoxic regions is a common characteristic of most solid tumors and is associated with radiation and chemotherapy resistance, as well as increased metastasis.[8] Levels of hypoxia also vary widely amongst patients, and are strongly associated with poor clinical outcome in several tumor types. Within individual tumors, hypoxia is spatially heterogeneous and often characterized by gradients of oxygenation extending from normal levels near the vasculature to near anoxia at the borders with necrosis. [9] Importantly, oxygen levels in some tumor regions also fluctuate over time due to transient changes in vessel perfusion. Areas of fluctuating hypoxia may be particularly important in driving poor prognosis, but at present are difficult to quantitate in patients.^[10] Current understanding of tumor hypoxia has been greatly facilitated by the availability of chemical probes constructed around a 2-nitroimidazole (2-NI) functionality such as pimonidazole (Pimo) and EF-5.[11] These probes are reduced by one electron reductases, such as PORs (p450 oxidoreductase), which act on a variety of nitroaryl compounds (Scheme 1b).[14] Studies suggest that upon reduction of the 2-NI to a 2-(hydroxy amino)imidazole (5), hydroxide is liberated and an electrophilic 2-(nitrenium)imidazole ion (6) is generated, which reacts irreversibly with free thiols in the

Scheme 1. a) a: MeLi (≈1 equiv), THF, 22 °C, 10 min; **b**: 3-chloro-1-propanol (1.0 equiv), THF, -192 °C, 2 h, 72%; **c**: p-nitrophenyl chloroformate (1.05 equiv), pyridine (2.1 equiv), THF, 22 °C, 2 h, 75%; **d**: methylbromoacetate (1.0 equiv), K_2CO_3 (1.5 equiv), tetrabutylammonium iodide (0.022 equiv), MeCN, reflux, 3 h, 60%; **e**: ethylenediamine (4.0 equiv), MeOH, 22 °C, 18 h, quant.; **f**: pyridine (3.0 equiv), MeOH, 22 °C, 2 h, 70%. b) Enzyme-catalyzed reduction of the 2-nitroimidazole functionality to produce the electrophilic protein-labeling nitrenium ion **6**.

hypoxic environment. This produces protein-2-(amino)imidazole adducts (7) effectively "tagging" the cell. Generation of the initial nitro radical (4) is a rapidly reversible process in the presence of molecular oxygen and thus generation of the active nitrenium labeling agent is slow under normoxic conditions. In the case of Pimo and EF-5, the detection is mediated by immunostaining, although similar 2-NI compounds have been developed which incorporate ¹⁸F to allow for detection through positron emission tomography. ^[12] Recently, hypoxia probes based on 2-NIs have been developed that implement alternative imaging modalities such as single-photon emission computed tomography, magnetic resonance imaging, or near-infrared fluorescence. ^[13]

To adapt the 2-NI functionality for MC detection we identified an organotellurium scaffold that was compact, amenable to high yielding synthesis, stable under physiological conditions, and relatively nontoxic. Reports of organotellurium toxicity have focused mainly on aryltelluroethers; however, the limited reports available suggest the more compact dialkyl telluroethers are less toxic.[15-17] Focusing on telluroethers, we opted to use a compact unsymmetric methyl alkyl telluride functionality (1, Scheme 1a) with a reactive hydroxy handle for further chemical manipulation. This group was accessed through treatment of elemental tellurium with methyllithium, followed by alkylation of the resultant nucleophilic methyltelluride salt by 3-chloropropan-1-ol (Scheme 1 a).[18] The resultant telluroether alcohol (1) was then treated with p-nitrophenyl chloroformate to afford the carbamylating reagent 2. The primary amine-bearing 2-NI scaffold (3) was then easily carbamylated to afford the final hypoxia probe, designated "Telox" (Scheme 1a). This compound is stable for months if stored in the dark as a solid and under an inert atmosphere at -20°C. At ambient light and oxygen levels the probe has a $t_{1/2} > 48 \text{ h}$ in solution (see the Supporting Information). Due to a downfield shift of the Te- CH_3 resonance in the ¹H NMR spectrum of the probe after prolonged exposure to ambient atmosphere, we hypothesize that the major degradation product upon exposure to atmospheric oxygen is the telluroxide species or a hydrate thereof.

The proliferative toxicity of Telox was measured in

HCT 116 cells by confluency analysis (Figure S1). This experiment suggested that proliferation was only mildly affected up to the maximum probe concentration evaluated (400 μM) under both normoxic (21 % O_2) and hypoxic (0.2 % O_2) conditions. An orthogonal assay was performed using the reduction of WST-1 as a metric for metabolic toxicity induced by the presence of Telox. In Jurkat cells, this experiment indicated a metabolic IC $_{50}$ of $200\pm20~\mu\text{M}$ (Figure S2).

As a surrogate for a POR enzyme, the sensitivity of Telox to enzyme-mediated reduction was

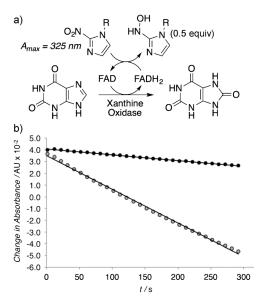


Figure 2. a) Xanthine oxidase catalyzed reduction of a generalized 2-nitroimidazole using xanthine as a source of electrons and FAD as a cofactor. b) Change in UV absorption over time at 325 nm for the anaerobic enzymatic reduction of the 2-nitroimidazole component of Pimonidazole (black circles) or Telox (grey circles). [2-NI] = 100 μm, [xanthine] = 500 μm, 0.2 units XO.

evaluated using bovine xanthine oxidase (XO; Figure 2a). [19] Following the loss of the 2-nitroimidazole absorption at 325 nm during XO-mediated oxidation of xanthine to uric acid demonstrated that under normoxia no reduction of Telox was observed (data not shown); however, under hypoxic conditions a clear loss of the 2-nitroimidazole occurred (Figure 2b). Interestingly the rate of reduction was higher for Telox than that for Pimo, one of the most commonly used hypoxia probes. These experiments suggest that the reduction potential and structure of Telox are compatible with enzymemediated reduction. Additionally, the telluroether functionality does not appear to have an inhibitory effect on the metal-containing active site of XO under these conditions.

Having confirmed that Telox could be enzymatically reduced in vitro, we evaluated the ability of the probe to label HCT116 cells under normoxic and hypoxic conditions (Figure 3). In an atmosphere that contained $<0.02\,\%$ $O_2,$ robust labeling (\approx 3.4-fold) was detected by ICP-MS analysis of whole cell pellets when cells were incubated for 3 h in media containing Telox (100 μM). A substantially lower tellurium concentration was detected under all other concentrations of O_2 , indicating that the labeling ability of Telox is indeed oxygen-dependent. This level of oxygen sensitivity is similar to that observed with EF5, which displays dramatic increases in labeling only below 0.1 % O_2 . Given the similar oxygen sensitivity, Telox should be useful as a MC-compatible surrogate for the widely used probes EF5 or Pimo.

Next, we investigated the ability of Telox to identify hypoxic cells in a mixture using MC. HCT116 cells were incubated under normoxic or anoxic conditions in the presence of Telox as detailed above. Following exposure to Telox the cells were washed and fixed. To correlate the cellular tellurium content with oxygen exposure, we used

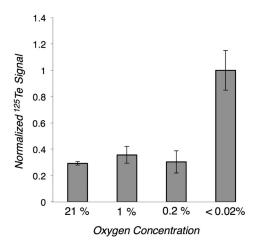


Figure 3. Traditional ICP-MS analysis of the tellurium content of whole HCT116 cell pellets as a function of cellular oxygen exposure. [Telox] = 100 μM, incubation time = 3 h.

metal-containing nucleic acid intercalators as internal standards for oxygen exposure. Cells incubated under normoxic conditions were treated with a rhodium-containing nucleic acid intercalator (Figure 1b), whereas cells incubated under anoxic conditions were treated with an iridium-containing nucleic acid intercalator (Figure 1b).^[7] Both samples were washed separately, combined, and injected onto a secondgeneration CyTOF instrument for MC analysis (Figure 4a). Generation of a density plot of event length versus ¹³⁰Te signal clearly indicated the presence of two distinct populations of cells (Figure 4b). Gating each population [high ¹³⁰Te (median \approx 120 cts) or low 130 Te (median \approx 30 cts)] and outputting 193 Ir versus ¹⁰³Rh density plots from these gates indicated that cells which contained a high relative amount of 130Te possessed a very high ¹⁹³Ir content and a low ¹⁰³Rh content (Figure 4c). Conversely, cells that produced a lower ¹³⁰Te signal contained far less 193Ir and a comparatively large amount of 103Rh (Figure 4 d). These results are consistent with expectations, as MC analysis was able to deconvolute the two cell populations to a high degree of resolution using tellurium content as a metric for the oxygen concentration to which cells were exposed. The non-zero 193 Ir signal observed in cells incubated under normoxic conditions was likely due to leeching of less tightly bound Ir intercalator between cell populations after mixing. This is supported by the observation that cells exposed only to the Rh intercalator and not mixed with Irstained cells exhibit a near-zero ¹⁹³Ir signal (Figure S3).

Finally, in an attempt to further confirm the mode of action, we evaluated the ability of Telox to compete for the same bioreductive pathway as the known hypoxia probe Pimo. Incubation of HCT116 cells in media that contained both probes at equal concentrations (100 μ M) reduced the tellurium labeling by 1.7-fold when compared to the signal in the absence of Pimo as indicated by CyTOF analysis (Figure 4e). This result suggests that these probes compete for at least some of the same reductase enzymes and that Telox should be a reliable surrogate for Pimo when studying cellular hypoxia.



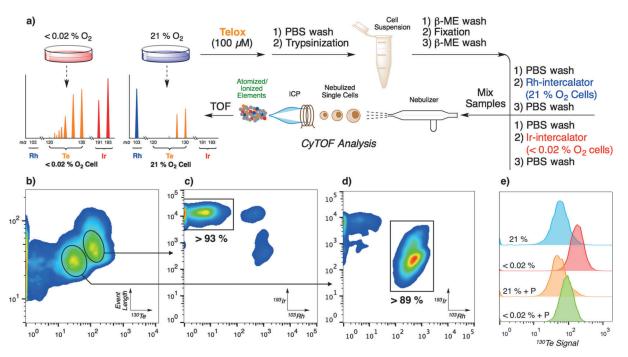


Figure 4. a) Schematic representation of Telox cell labeling for analysis by mass cytometry on a second-generation CyTOF instrument. b) Density map of signal event length vs. 130 Te signal (arbitrary units). c) Density plot output from the top-right gate in (b) of the 193 Ir signal versus the 103 Rh signal. More than 93% of the detected events fall in the square gate. d) Density plot output from the bottom-left gate in (b) of the 193 Ir signal versus the 103 Rh signal. More than 89% of the detected events fall in the square gate. e) Population histograms of cell 130 Te content. Oxygen concentrations are listed as numerical percentages, P = Pimonidazole (100 μ m). Blue and red histograms are Pimonidazole-negative controls. Orange and green histograms are the Pimonidazole-positive competition experiments. Note: warmer colors in density plots indicate higher cell population density.

Moving forward, we propose that the MC-compatible hypoxia probe developed in this report will be of value for the study of complex relationships between hypoxia and tumor biochemistry through the development of highly parameter assays in combination with MaxPar reagents. Furthermore, experiments can be envisioned in which isotopically pure tellurium probes are synthesized and employed in pulsechase-type experiments to identify the effects that various xenobiotic agents have on the hypoxic response. Our probe is superior to fluorophore-based molecules for this purpose because isotopologs of Telox would be structurally identical and thus have identical pharmacokinetics/dynamics, unlike 2-NIs conjugated to structurally diverse fluorophores. Additionally, the tellurium-containing p-nitrophenyl carbonate ester 2 is a versatile reagent for carbamylation of amines, thus presenting a general methodology for accessing new tellurium-containing activity-based probes to be used with MC.

Received: May 13, 2014 Revised: July 9, 2014

Published online: September 3, 2014

Keywords: activity-based probes \cdot mass cytometry \cdot multiparameter analysis \cdot organotellurium compounds \cdot tumor hypoxia

- [2] O. Ornatsky, D. Bandura, V. Baranov, M. Nitz, M. A. Winnik, S. Tanner, J. Immunol. Methods 2010, 361, 1.
- [3] O. Ornatsky, R. Kinach, D. R. Bandura, X. Lou, S. D. Tanner, V. I. Baranov, M. Nitz, M. A. Winnik, J. Anal. At. Spectrom. 2008, 23, 463
- [4] S. C. Bendall, E. F. Simonds, P. Qiu, E. D. Amir, P. O. Krutzik, R. Finck, R. V. Bruggner, R. Melamed, A. Trejo, O. I. Ornatsky, R. S. Balderas, S. K. Plevritis, K. Sachs, D. Pe'er, S. D. Tanner, G. P. Nolan, *Science* 2011, 332, 687.
- [5] a) C. Giesen, H. A. O. Wang, D. Schapiro, N. Zivanovic, A. Jacobs, B. Hattendorf, P. J. Schüffler, D. Grolimund, J. M. Buhmann, S. Brandt, Z. Varga, P. J. Wild, D. Günther, B. Bodenmiller, *Nat. Methods* 2014, 11, 417; b) M. Angelo, S. C. Bendall, R. Finck, M. B. Hale, C. Hitzman, A. D. Borowsky, R. M. Levenson, J. B. Lowe, S. D. Liu, S. Zhao, Y. Natkunam, G. P. Nolan, *Nat. Med.* 2014, 20, 436.
- [6] X. Lou, G. Zhang, I. Herrera, R. Kinach, O. Ornatsky, V. Baranov, M. Nitz, M. A. Winnik, Angew. Chem. Int. Ed. 2007, 46, 6111; Angew. Chem. 2007, 119, 6223.
- [7] a) O. I. Ornatsky, X. Lou, M. Nitz, S. Schäfer, W. S. Sheldrick, V. I. Baranov, D. R. Bandura, S. D. Tanner, *Anal. Chem.* 2008, 80, 2539; b) G. K. Behbehani, S. C. Bendall, M. R. Clutter, W. J. Fantl, G. P. Nolan, *Cytometry Part A* 2012, 81, 552.
- [8] M. W. Dewhirst, Y. Cao, B. Moeller, Nat. Rev. Cancer 2008, 7, 25.
- [9] M. Nordsmark, S. M. Bentzen, V. Rudat, D. Brizel, E. Lartigau, P. Stadler, A. Becker, M. Adam, M. Molls, J. Dunst, D. J. Terris, J. Overgaard, *Radiother. Oncol.* 2005, 77, 18.
- [10] J. M. Brown, W. R. Wilson, Nat. Rev. Cancer 2004, 4, 437.
- [11] a) W. R. Wilson, M. P. Hay, Nat. Rev. Cancer 2011, 11, 393; b) S. Kizaka-Kondoh, H. Konse-Nagasawa, Cancer Sci. 2009, 11, 1366; c) M. A. Varia, D. P. Calkins-Adams, L. H. Rinker, A. S. Kennedy, D. B. Novotny, W. C. Fowler Jr., J. A. Raleigh, Gynecol. Oncol. 1998, 71, 270.

S. C. Bendall, G. P. Nolan, M. Roederer, P. K. Chattopadhyay, Trends Immunol. 2012, 33, 323.



- [12] L. S. Ziemer, S. M. Evans, A. V. Kachur, A. L. Shuman, C. A. Cardi, W. T. Jenkins, J. S. Karp, A. Alavi, W. R. Dolbier, Jr., C. J. Koch, *Eur. J. Nucl. Med.* **2003**, *30*, 259.
- [13] a) F. A. Rojas-Quijano, G. Tircsó, E. T. Benyó, Z. Baranyai, H. T. Hoang, F. K. Kálmán, P. K. Gulaka, V. D. Kodibagkar, S. Aime, Z. Kovács, A. D. Sherry, *Chem. Eur. J.* 2012, 18, 9669; b) C. Hsia, F. Huang, G. Hung, L. Shen, C. Chen, H. Wang, *Appl. Radiat. Isot.* 2011, 69, 649; c) K. Okuda, Y. Okabe, T. Kadonosono, T. Ueno, B. G. M. Youssif, S. Kizaka-Kondoh, H. Nagasawa, *Bioconjugate Chem.* 2012, 23, 324.
- [14] a) R. A. McClelland, R. Panicucci, A. M. Rauth, J. Am. Chem. Soc. 1985, 107, 1762; b) J. L. Bolton, R. A. McClelland, J. Am. Chem. Soc. 1989, 111, 8172; c) R. J. Hodgkiss, Anti-Cancer Drug

- Des. 1998, 13, 687; d) D. C. Heimbrook, A. C. Sartorelli, Mol. Pharmacol. 1985, 29, 168.
- [15] L. A. Ba, M. Döring, V. Jamier, C. Jacob, Org. Biomol. Chem. 2010, 8, 4203.
- [16] L. Engman, N. Al-Maharik, M. McNaughton, A. Birmingham, G. Powis, *Bioorg. Med. Chem.* 2003, 11, 5091.
- [17] L. Engman, T. Kanda, A. Gallegos, R. Williams, G. Powis, Anti-Cancer Drug Des. 2000, 15, 323.
- [18] A. J. Barton, W. Levason, G. Reid, A. J. Ward, *Organometallics* 2001, 20, 3644.
- [19] E. D. Clarke, K. H. Goulding, P. Wardman, *Biochem. Pharma-col.* 1982, 31, 3237.
- [20] S. M. Evans, C. J. Koch, Cancer Lett. 2003, 195, 1.